## The Crystal Structure of [(Prop-cis-2-enylphenyl)diphenylphosphine]tetracarbonylmolybdenum(0)

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THE catalytic isomerisation of 2-allylphenyldiphenylphosphine ( $Ph_2PC_6H_4$ ·CH<sub>2</sub>·CH=CH<sub>2</sub>) to (prop-2-enylphenyl)diphenylphosphine ( $Ph_2PC_6H_4$ -CH=CH·CH<sub>3</sub>), pp, has been shown<sup>1,2</sup> to occur on formation of a complex with certain metals, Pt, Cr, Mo, and W. For M = Cr, Mo, and W, the intermediate complexes, M(CO)<sub>4</sub>pp were isolated; investigation by n.m.r. and i.r. spectroscopy indicated that they were isostructural, with the strongest M-C bonding in the tungsten compound. It appeared that the isomerisation had taken place during the formation of the metal complex to give an olefin which was  $\pi$ -bonded to the metal.

We have determined the crystal structure of the

complex Mo(CO)<sub>4</sub> cis-pp from three-dimensional X-ray diffraction measurements at  $150-155^{\circ}$ K. In a triclinic unit cell having a = 10.53, b = 13.26, c = 9.36 Å,  $\alpha = 100.3^{\circ}$ ,  $\beta = 91.0^{\circ}$ ,  $\gamma = 118.2^{\circ}$  and space group  $P\overline{1}$  there are two molecules. The molybdenum and phosphorus atoms were located by Patterson syntheses and the light atoms, excluding hydrogen, from electron density difference syntheses. At present R = 0.15 for 3512 observations and refinement is being continued.

The stereochemistry is shown in the Figure. It establishes that  $deduced^{1,2}$  from i.r. and n.m.r. spectra. In particular the *cis* configuration of the olefinic bond in the catalytic intermediate is

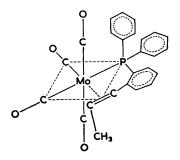


FIGURE. One molecule of [(prop-cis-2-enylphenyl)diphenylphosphine]tetracarbonylmolybdenum(0). The drawing is somewhat idealised; in a projection with the other atoms resolved, as shown, the two unsubstituted phenyl groups overlap one another.

confirmed. The preferential formation of the *cis*isomer which has a higher energy than the *trans*isomer is remarkable because the stable complex  $Mo(CO)_4$ -*trans*-pp can be made directly from the *trans*-ligand, and experiments with models show that it could have a structure differing from that in the Figure merely by the C-CH<sub>3</sub> bending "upwards" instead of "downwards."

Detailed discussion of bond lengths will be deferred until refinement is complete but some features of interest are unlikely to change significantly. If the mid-point of the olefinic bond is considered as one ligand, all the angles subtended at the molybdenum atom are  $90^{\circ}$  or  $180^{\circ}$  (within  $3^{\circ}$ ). Both olefinic carbon atoms lie in the plane (A) defined by the molybdenum and phosphorus atoms and two carbonyl groups. The bond length in the olefin is  $1.40 \pm 0.02$  Å, significantly longer than a C=C double bond, while the C-C=C angles are 126° and 124°, consistent with  $sp^2$  hybridisation without additional p character. The molybdenum to mid-point direction, shown by a broken line in the Figure, is not normal to the C(methyl)-C=C-C-C(phenyl) plane, the substituents are bent away from the molybdenum so that the normal to this plane makes an angle of  $107^{\circ}$  with the normal to the plane (A). This may be merely a steric effect but it is consistent with distortion arising from the presence of electrons in the antibonding  $\pi^*$ -orbital of the co-ordinated olefin.

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<sup>1</sup> M. A. Bennett, L. V. Interrante, and R. S. Nyholm, Z. Naturforsch., 1965, 20b, 633. <sup>2</sup> L. V. Interrante, M. A. Bennett, and R. S. Nyholm, Inorg. Chem., 1966, 5, 2212.